2023

4th Semester Examination CHEMISTRY (Honours)

Paper: C 10-T

(Organic Chemistry-IV)

[CBCS]

Full Marks: 40

Time: Two Hours

The figures in the margin indicate full marks. Candidates are required to give their answers in their own words as far as practicable.

Group - A

Answer any five questions:

 $2 \times 5 = 1.0$

- 1. Explain the following:
 - (i) Ethylene fails to show any IR signals around 1600 cm⁻¹.
 - (ii) Can D₂O be used as a solvent in ¹H-NMR studies?
- 2. Write a note on symmetry restriction in electronic transition.

- 3. How can you distinguish chemically between N-methylaniline and N, N-dimethylaniline?
- 4. In the Arndt-Ester synthesis two equivalents of diazomethane is used. Explain the statement showing mechanism of the reaction.
- 5. Alkaline hydrolysis of benzonitrile affords the salt of an acid but in presence of H_2O_2 an amide is formed Explain.
- 6. Transform the following:

- 7. How you can distinguish between the following pairs:
 - (i) CH₃CH₂Br and CH₃CH₂OH (ordinary grade) by NMR.
 - (ii) Methylbenzoate and Phenylacetate by IR.

8. Convert the following:

$H_3C COCH_2CH_2CO_2Et \rightarrow CH_3CH(OH)CH_2CH_2CO_2Et$ $CH_3COCH_2CH_2CH_2OH$

Group - B

Answer any *four* questions : $5 \times 4 = 20$

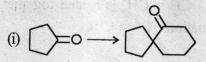
9. (a) Complete the following reaction scheme with plausible mechanism.

$$PhCH_{2}NH_{2} \xrightarrow{\text{(i) CICOOEt}} (A) \xrightarrow{NaOH/H_{2}O} (B)$$

- (b) How can IR spectroscopy distinguish between 1-hexyne, 2-hexyne, and 3-hexyne?
- (a) An organic compound (S) C₈H₉ON on treatment with H₂SO₄ isomerizes to (T) which on hydrolysis furnishes aniline and acetic acid. What are (S) and (T)? Explain the above fact and show mechanism of isomerization step only.
 - (b) How *o*-nitroaniline can be distinguished from *p*-nitroaniline by UV-spectroscopy?
- 11. (a) Write the structure of all possible products when 1:1 mixture of PhOCH₂ CH = C^*H_2 and PhOCH₂ CH = CH_2 is heated together. Explain their formation. ($C^* = C^{14}$).

P.T.O.

- (b) Why do acetylenic protons resonate at upfield region with respect to ethylenic protons though acetylenic hydrogens are more acidic than ethylenic hydrogens? Explain.
- 12. (a) Compare the rate of formation of aniline from fluorobenzene and bromobenzene in presence of $NaNH_2$ and liq. NH_3 . Cite suitable experimental evidence.
 - (b) Which has a greater chemical shift for the OH proton, the ¹H NMR spectrum of pure ethanol or the ¹H NMR spectrum of ethanol dissolved in CH₂Cl₂? Give your answer with explanation. 2
- 13. (a) Explain with mechanism the rate(s) of nitration of nitrobenzene and pentadeuteronitrobenzene under similar reaction conditions.
 - (b) How many peaks do you expect for nitrobenzene in its ¹H-NMR spectrum? Draw a rough sketch for the ¹H-NMR spectrum of nitrobenzene assigning the protons in the diagram. (Actual δ values and meta-coupling may be ignored).
- 14. (a) If two signals differ by 90 Hz in a 300 MHz spectrometer, by how much do they differ in a 500 MHz spectrometer?
 - (b) Convert: 2+2



(ii)
$$\longrightarrow$$
 PH OH NMe₂

Group - C

Answer any one question:

 $10 \times 1 = 10$

15. (a) Give the retrosynthetic pathways and one efficient synthetic method for the following compound. 3×2

(b) Write down the structure of the products A, B, C and D 4

$$\begin{array}{c|c} & & & \\ \hline & &$$

P.T.O.

16. (a) Predict the major product only and write mechanism to show their formation. 2×4

(iii)
$$NH-NH- A$$
 A

(b) Which member of the following pair will undergo Dienone-Phenol rearrangement more rapidly and why?